

Periodic Graphics

A collaboration between C&EN and
Andy Brunning, author of the popular
graphics blog *Compound Interest*

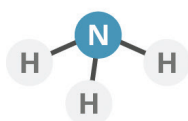
More
online

To see more of
Brunning's work, go to
compoundchem.com.
To see all of C&EN's
Periodic Graphics,
visit **cenm.ag/
periodicgraphics**.

ENVIRONMENTAL IMPACT OF INDUSTRIAL REACTIONS

The chemical industry accounts for about 10% of the world's energy demand and 7% of its greenhouse emissions. Here we take a look at the top 5 chemical products responsible.

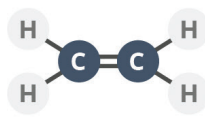
1 AMMONIA



Synthesis
Haber-Bosch process

Major uses
Fertilizers, medicines,
and cleaning products

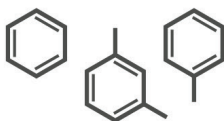
2 ETHYLENE



Synthesis
Cracking of long-chain
hydrocarbons

Major uses
Making polyethylene

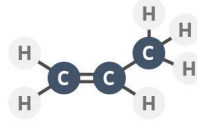
3 AROMATICS



Synthesis
Catalytic reforming of
naphtha

Major uses
Solvents and reagents in
chemical reactions

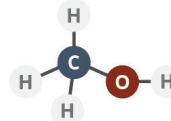
4 PROPYLENE



Synthesis
Cracking of long-chain
hydrocarbons

Major uses
Making polypropylene

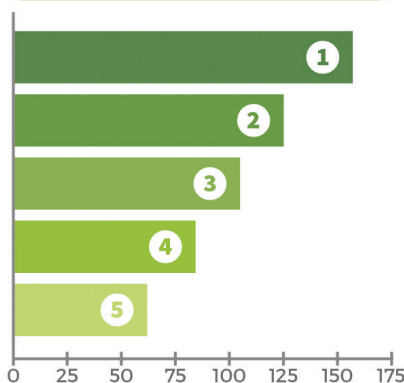
5 METHANOL



Synthesis
Reactions with CO, CO₂,
and H₂

Major uses
Making other chemicals

PRODUCTION VOLUMES

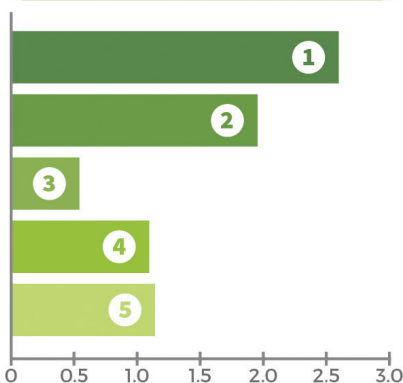


MEGA-METRIC TONS/YEAR



1 Mt: Equivalent to the mass of
nearly 5,000 Statues of Liberty

ENERGY CONSUMPTION

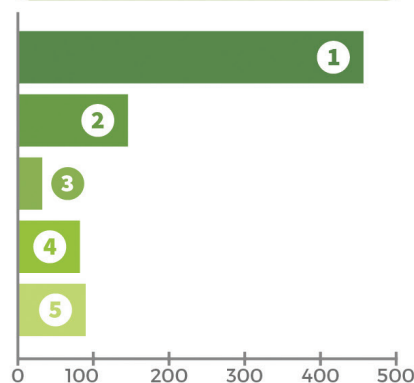


EXAJOULES/YEAR



1 EJ: Equivalent to
the energy from about 174
million oil barrels

GREENHOUSE GAS EMISSIONS



Mt of CO₂ EQUIVALENT/YEAR



1 Mt CO₂ eq: Equivalent to 25%
of the CO₂ emitted from a coal
power plant per year

Source: DECHEMA, 2010. For ethylene and propylene, figures are representative of the steam cracking process.



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